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How to Find the Percent

Composition by Mass for

$\text{Al}_2(\text{SO}_4)_3$ (Aluminum sulfate)

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composition of all elements in
these compounds Show all work 1
ammonium sulfite N 24 1 H 6 9 S

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27 6 O 41 3 2 aluminum acetate Al
13 2 C 35 3 The percent
composition by mass of a
compound represents the percent
that each element in a compound
contributes to the total mass of the
compound A chemist often
compares the percent composition
of ...

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Percentage Composition

Worksheet – Answer Key Give the
% composition of all elements in
these compounds. Show all work!

_____ 1) ammonium sulfite % N
24.1 % H 6.9 % S 27.6 % O 41.3

_____ 2) aluminum acetate % Al
13.2 % C 35.3

Percentage Composition

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